

M.G.V.C ART'S COMMERCE AND SCIENCE COLLEGE, MUDDEBIHAL-586212
DEPARTMENT OF CHEMISTRY
REMEDIAL ENRICHMENT CLASSES 2024-25
LIST OF B.Sc. I SEMESTER STUDENTS

Sl.No.	Reg No.	Name of the Student	Gender	Subject Combination
1	U15NU24S0001	Mahant Malagimani	M	CBZ
2	U15NU24S0002	Sahana Patil	F	PCM
3	U15NU24S0003	SANGAMMA R MANAGULI	F	PCM
4	U15NU24S0004	Shruti Hanamantray Singnalli	F	CBZ
5	U15NU24S0005	Srushti B Naduvanamath	F	PCM
6	U15NU24S0006	SARASWATI BAGEWADI	F	PCM
7	U15NU24S0007	SWATI DODAMANI	F	PCM
8	U15NU24S0008	Mahajabina J Bavura	F	PCM
9	U15NU24S0009	Boramma	F	CBZ
10	U15NU24S0010	Madhu	F	CBZ
11	U15NU24S0011	Tasneem Sagar	F	CBZ
12	U15NU24S0012	Siddu Basavaraj Hullur	M	PCM
13	U15NU24S0013	Aishwarya	F	CBZ
14	U15NU24S0014	Devamma	F	CBZ
15	U15NU24S0015	Pooja Panchangamatha	F	PCM
16	U15NU24S0016	Vidyashree Naikodi	F	CBZ
17	U15NU24S0017	Swapna S Malagaladinni	F	CBZ
18	U15NU24S0018	Gayatri	F	CBZ
19	U15NU24S0019	Devamma Avanna Nalatawad	F	PCM
20	U15NU24S0020	Prashant Beerappa Niralagi	M	PCM
21	U15NU24S0021	Bapugouda K Patil	M	CBZ
22	U15NU24S0023	Gaddemma B Desai	F	PCM
23	U15NU24S0024	Shreekant R Kondaguli	M	PCM
24	U15NU24S0025	Sahana Gurappa Takkalaki	F	CBZ
25	U15NU24S0026	Shilpa K Kuri	F	PCM
26	U15NU24S0027	Bhagyashree KChalawadi	F	PCM
27	U15NU24S0028	Pratish Yamanappa Malaghan	M	CBZ
28	U15NU24S0029	Yusuf H Turakanageri	M	PCM
29	U15NU24S0030	Hanamantrayagouda Patil	M	PCM
30	U15NU24S0031	Suresh C	M	CBZ
31	U15NU24S0032	Najmunnissa Ghati	F	CBZ
32	U15NU24S0033	Shahista Donur	F	CBZ
33	U15NU24S0034	Sakshi Shrishail Ukkali	F	CBZ
34	U15NU24S0035	Pallavi Areppa Talageri	F	CBZ
35	U15NU24S0036	Prabhu Mallappa Biradar	M	PCM
36	U15NU24S0037	Rakshita Basavaraj Sulibhavi	F	CBZ
37	U15NU24S0038	Rahul Rathod	M	CBZ

38	U15NU24S0039	Dawalsab Sankanlal	M	CBZ
39	U15NU24S0042	Roopa Mallikarjun Channa	F	CBZ
40	U15NU24S0043	Ganesh kolar	M	CBZ
41	U15NU24S0044	Jayalaxmi S Biradar	F	PCM
42	U15NU24S0045	Shivaleela Masuti	F	CBZ
43	U15NU24S0046	Maruti Naragal	M	CBZ
44	U15NU24S0047	Swati Biradar	F	PCM
45	U15NU24S0048	Tahreen Bagalkot	F	CBZ
46	U15NU24S0049	Suresh Policepatil	M	PCM
47	U15NU24S0050	Sahana Chidambar Desai	F	PCM
48	U15NU24S0051	Shanta Sanganagouda Patil	F	PCM
49	U15NU24S0052	Shabbir Imamasa Nadaf	M	CBZ
50	U15NU24S0053	Vikas	M	PCM
51	U15NU24S0054	Varshita Shillina	F	CBZ
52	U15NU24S0055	Jayashree Somalappa Jadhav	F	CBZ
53	U15NU24S0057	Surammadevi M Meti	F	CBZ
54	U15NU24S0059	Aishwarya Shrishail Hatti	F	PCM
55	U15NU24S0060	Afreen Ramajanasaba Nadaf	F	PCM
56	U15NU24S0061	Laxman Parasappa Kolakar	M	PCM
57	U15NU24S0062	Purushottam PHavaladar	M	CBZ
58	U15NU24S0063	BHEEMARAY SUNAGAR	M	PCM
59	U15NU24S0064	Kavita	F	PCM
60	U15NU24S0065	Nandini Havaladar	F	CBZ
61	U15NU24S0066	Susmita Guruvini	F	CBZ
62	U15NU24S0068	Deepa Ningappa Arabi	F	PCM
63	U15NU24S0069	Parashuram HDharenavar	M	CBZ
64	U15NU24S0070	Srushthi Siddanna Waggar	F	PCM
65	U15NU24S0071	Shivatmaja Ashok Tondihal	M	CBZ
66	U15NU24S0072	Shifa Bagwan	F	PCM
67	U15NU24S0073	Sushmita Onarotti	F	PCM
68	U15NU24S0074	Bonashri Basayya Mathapati	F	PCM
69	U15NU24S0075	SAGAR BIDARAKUNDI	M	PCM
70	U15NU24S0076	Misabhajahan M Naikwadi	F	CBZ
71	U15NU24S0077	SIDIYA M NAIKODI	F	CBZ
72	U15NU24S0078	Vijayalaxmi	F	PCM
73	U15NU24S0079	Sadika Allampur	F	PCM
74	U15NU24S0081	LAKSHMI NAGARABETTA	F	CBZ
75	U15NU24S0082	Madhushree	F	PCM
76	U15NU24S0083	Shweta Laxman Byalyal	F	PCM
77	U15NU24S0084	Siddanagouda SPatil	M	CBZ
78	U15NU24S0085	Anuradha Badiger	F	PCM
79	U15NU24S0086	Prabhuling	M	CBZ
80	U15NU24S0087	Mallappa	M	CBZ
81	U15NU24S0088	Laxmi Huligappa Madar	F	CBZ

82	U15NU24S0089	Shiddanagouda HSankanal	M	CBZ
83	U15NU24S0090	Shreeshail Ramesh Nagaral	M	PCM
84	U15NU24S0091	Prerana Kumbhar	F	PCM
85	U15NU24S0092	PRADEEP HULAGANNI	M	PCM
86	U15NU24S0093	ARAVINDA R GASTIGAR	M	PCM
87	U15NU24S0094	Ananya R Patil	F	CBZ
88	U15NU24S0095	Swapna Chalawadi	F	CBZ
89	U15NU24S0096	Hinakousar Ramjansab Nadaf	F	PCM
90	U15NU24S0097	Muthanna Talawar	M	PCM
91	U15NU24S0098	Rohit Pandappa Nayak	M	CBZ
92	U15NU24S0100	Prashant Mural	M	PCM
93	U15NU24S0101	ANJUMA Y NADAF	F	CBZ
94	U15NU24S0102	Bhagyashree N Bapparagi	F	CBZ
95	U15NU24S0103	Vidya	F	CBZ
96	U15NU24S0104	ABDUL LATIF K NADAF	M	PCM
97	U15NU24S0105	Najiya Kashimasa Walikar	F	PCM
98	U15NU24S0106	SAMEER NADAF	M	CBZ
99	U15NU24S0107	Arati Pawar	F	CBZ
100	U15NU24S0108	Madhumati Shivappa Chouri	F	PCM
101	U15NU24S0109	Arshiya Bagalkot	F	CBZ
102	U15NU24S0110	Pavitra Jummanna Somanal	F	PCM
103	U15NU24S0111	Ramanagoud Byakod	M	CBZ
104	U15NU24S0112	Akshata Arjun Koppad	F	CBZ
105	U15NU24S0113	RAGHAVENDRA	M	CBZ
106	U15NU24S0114	Prajwal S Malipatil	M	PCM
107	U15NU24S0115	SANGAMESH WADAWADAGI	M	PCM
108	U15NU24S0116	Jayashree Krishnappa Lamani	F	CBZ
109	U15NU24S0117	Gangadhar	M	PCM
110	U15NU24S0118	Basavant Vijapur	M	CBZ
111	U15NU24S0119	Yuvaraj Ningappa Undigeri	M	PCM
112	U15NU24S0120	Saniya Myageri	F	CBZ
113	U15NU24S0121	Santosh Sangam	M	PCM
114	U15NU24S0122	Saniya Naj A Salwadagi	F	CBZ
115	U15NU24S0123	Amruta Bavikatti	F	PCM
116	U15NU24S0124	Prabhagya	F	CBZ
117	U15NU24S0125	Sahana Andeli	F	PCM
118	U15NU24S0126	Shreekant Chavhan	M	CBZ
119	U15NU24S0127	Sushma M Sangam	F	CBZ
120	U15NU24S0129	Parveen Nadaf	F	PCM
121	U15NU24S0130	Susma Nayakamakkala	F	PCM
122	U15NU24S0131	Madivalappa Suresh Vijapur	M	CBZ
123	U15NU24S0133	Inzamam Hadagali	M	PCM
124	U15NU24S0134	Chetan Ramesh Kanamadi	M	PCM
125	U15NU24S0135	Yallamma	F	PCM
126	U15NU24S0136	Prema	F	PCM


 Head, Dept. of Chemistry,
 M.G.V.C. Arts, Com. & Science College
 MUDDERHAL - 580212.

M.G.V.C ART'S COMMERCE AND SCIENCE COLLEGE, MUDDEBIHAL-586212
DEPARTMENT OF CHEMISTRY
ENTRY LEVEL RESULT OF B.Sc-I SEMESTER STUDENTS 2024-25

SL.No.	Reg. No.	Name of the Student	PUC-II Year	
			Marks	%
1	U15NU24S0001	Mahant Malagimani	333	55.5
2	U15NU24S0002	Sahana Patil	435	72.5
3	U15NU24S0003	SANGAMMA R MANAGULI	548	91.3
4	U15NU24S0004	Shruti Hanamantray Singnalli	524	87.3
5	U15NU24S0005	Srushti B Naduvinamath	580	96.7
6	U15NU24S0006	SARASWATI BAGEWADI	449	74.8
7	U15NU24S0007	SWATI DODAMANI	532	88.7
8	U15NU24S0008	Mahajabina J Bavura	453	75.5
9	U15NU24S0009	Boramma	490	81.7
10	U15NU24S0010	Madhu	509	84.8
11	U15NU24S0011	Tasneem Sagar	527	87.8
12	U15NU24S0012	Siddu Basavaraj Hullur	430	71.7
13	U15NU24S0013	Aishwarya	467	77.8
14	U15NU24S0014	Devamma	466	77.7
15	U15NU24S0015	Pooja Panchangamatha	566	94.3
16	U15NU24S0016	Vidyashree Naikodi	545	90.8
17	U15NU24S0017	Swapna S Malagaladinni	504	84.0
18	U15NU24S0018	Gayatri	456	76.0
19	U15NU24S0019	Devamma Avanna Nalatawad	561	93.5
20	U15NU24S0020	Prashant Beerappa Niralagi	462	77.0
21	U15NU24S0021	Bapugouda K Patil	460	76.7
22	U15NU24S0023	Gaddemma B Desai	486	81.0
23	U15NU24S0024	Shreekant R Kondaguli	573	95.5
24	U15NU24S0025	Sahana Gurappa Takkalaki	519	86.5
25	U15NU24S0026	Shilpa K Kuri	557	92.8
26	U15NU24S0027	Bhagyashree KChalawadi	512	85.3
27	U15NU24S0028	Pratish Yamanappa Malaghan	498	83.0
28	U15NU24S0029	Yusuf H Turakanageri	483	80.5
29	U15NU24S0030	Hanamantrayagouda Patil	492	82.0
30	U15NU24S0031	Suresh C	386	64.3
31	U15NU24S0032	Najmunnissa Ghati	464	77.3
32	U15NU24S0033	Shahista Donur	433	72.2
33	U15NU24S0034	Sakshi Shrishail Ukkali	479	79.8
34	U15NU24S0035	Pallavi Areppa Talageri	516	86.0
35	U15NU24S0036	Prabhu Mallappa Biradar	539	89.8
36	U15NU24S0037	Rakshita Basavaraj Sulibhavi	488	81.3
37	U15NU24S0038	Rahul Rathod	507	84.5
38	U15NU24S0039	Dawalsab Sankanai	441	73.5

39	U15NU24S0042	Roopa Mallikarjun Channa	583	97.2
40	U15NU24S0043	Ganesh kolar	428	71.3
41	U15NU24S0044	Jayalaxmi S Biradar	577	96.2
42	U15NU24S0045	Shivaleela Masuti	479	79.8
43	U15NU24S0046	Maruti Naragal	547	91.2
44	U15NU24S0047	Swati Biradar	506	84.3
45	U15NU24S0048	Tahreen Bagalkot	364	60.7
46	U15NU24S0049	Suresh Policepatil	496	82.7
47	U15NU24S0050	Sahana Chidambar Desai	501	83.5
48	U15NU24S0051	Shanta Sanganagouda Patil	533	88.8
49	U15NU24S0052	Shabbir Imamasa Nadaf	535	89.2
50	U15NU24S0053	Vikas	450	75.0
51	U15NU24S0054	Varshita Shillina	448	74.7
52	U15NU24S0055	Jayashree Somalappa Jadhav	492	82.0
53	U15NU24S0057	Surammadevi M Meti	532	88.7
54	U15NU24S0059	Aishwarya Shrishail Hatti	472	78.7
55	U15NU24S0060	Afreen Ramajanasaba Nadaf	454	75.7
56	U15NU24S0061	Laxman Parasappa Kolakar	501	83.5
57	U15NU24S0062	Purushottam PHavaladar	386	63.3
58	U15NU24S0063	BHEEMARAY SUNAGAR	462	77.0
59	U15NU24S0064	Kavita	556	92.7
60	U15NU24S0065	Nandini Havaladar	558	93.0
61	U15NU24S0066	Susmita Guruvin	520	86.7
62	U15NU24S0044	Jayalaxmi S Biradar	504	67.5
63	U15NU24S0069	Parashuram H Dharennavar	348	58.0
64	U15NU24S0070	Srushthi Siddanna Waggar	570	95.0
65	U15NU24S0071	Shivatmaja Ashok Tondihal	468	78.0
66	U15NU24S0072	Shifa Bagwan	531	88.5
67	U15NU24S0073	Sushmita Onarotti	516	86.0
68	U15NU24S0074	Bonashri Basayya Mathapati	522	87.0
69	U15NU24S0075	SAGAR BIDARAKUNDI	434	72.3
70	U15NU24S0076	Misabhajahan M Naikwadi	481	80.2
71	U15NU24S0077	SIDIYA M NAIKODI	413	68.8
72	U15NU24S0078	Vijayalaxmi	490	81.7
73	U15NU24S0079	Sadika Allampur	478	79.7
74	U15NU24S0081	LAKSHMI NAGARABETTA	547	91.2
75	U15NU24S0082	Madhushree	500	83.3
76	U15NU24S0083	Shweta Laxman Byalyal	451	75.2
77	U15NU24S0084	Siddanagouda SPatil	485	80.8
78	U15NU24S0085	Anuradha Badiger	546	91.0
79	U15NU24S0086	Prabhuling	555	92.5
80	U15NU24S0087	Mallappa	521	86.8
81	U15NU24S0088	Laxmi Huligappa Madar	422	70.3
82	U15NU24S0089	Shiddanagouda HSankanal	501	83.5

83	U15NU24S0090	Shreeshail Ramesh Nagaral	540	90.0
84	U15NU24S0091	Prerana Kumbar	473	78.8
85	U15NU24S0092	PRADEEP HULAGANNI	520	86.7
86	U15NU24S0093	ARAVINDA R GASTIGAR	484	80.7
87	U15NU24S0094	Ananya R Patil	481	80.2
88	U15NU24S0095	Swapna Chalawadi	491	81.8
89	U15NU24S0096	Hinakousar Ramjansab Nadaf	440	73.3
90	U15NU24S0097	Muthanna Talawar	482	80.3
91	U15NU24S0098	Rohit Pandappa Nayak	417	69.5
92	U15NU24S0100	Prashant Mural	564	94.0
93	U15NU24S0101	ANJUMA Y NADAF	470	78.3
94	U15NU24S0102	Bhagyashree N Bapparagi	521	86.8
95	U15NU24S0103	Vidya	529	88.2
96	U15NU24S0104	ABDUL LATIF K NADAF	355	59.2
97	U15NU24S0105	Najiya Kashimasa Walikar	481	80.2
98	U15NU24S0106	SAMEER NADAF	435	72.5
99	U15NU24S0107	Arati Pawar	539	89.8
100	U15NU24S0108	Madhumati Shivappa Chouri	574	95.7
101	U15NU24S0109	Arshiya Bagalkot	481	80.2
102	U15NU24S0110	Pavitra Jummanna Somanal	548	91.3
103	U15NU24S0111	Ramanagoud Byakod	463	77.2
104	U15NU24S0112	Akshata Arjun Koppad	496	82.7
105	U15NU24S0113	RAGHAVENDRA	449	74.8
106	U15NU24S0114	Prajwal S Malipatil	498	83.0
107	U15NU24S0115	SANGAMESH WADAWADAGI	428	71.3
108	U15NU24S0116	Jayashree Krishnappa Lamani	541	90.2
109	U15NU24S0117	Gangadhar	554	92.3
110	U15NU24S0118	Basavant Vijapur	556	92.7
111	U15NU24S0119	Yuvaraj Ningappa Undigeri	502	83.7
112	U15NU24S0120	Saniya Myageri	501	83.5
113	U15NU24S0121	Santosh Sangam	438	73.0
114	U15NU24S0122	Saniya Naj A Salwadagi	538	89.7
115	U15NU24S0123	Amruta Bavikatti	546	91.0
116	U15NU24S0124	Prabhagya	447	74.5
117	U15NU24S0125	Sahana Andeli	523	87.2
118	U15NU24S0126	Shreekant Chavhan	444	74.0
119	U15NU24S0127	Sushma M Sangam	495	82.5
120	U15NU24S0129	Parveen Nadaf	475	79.2
121	U15NU24S0130	Susma Nayakamakkala	463	77.2
122	U15NU24S0131	Madivalappa Suresh Vijapur	476	79.3
123	U15NU24S0133	Inzamam Hadagali	486	81.0
124	U15NU24S0134	Chetan Ramesh Kanamadi	536	89.3
125	U15NU24S0135	Yallamma	497	82.8
126	U15NU24S0136	Prema	493	82.2


 Head, Dept. of Chemistry,
 M.G.V.C. Arts, Com. & Science College
 MUDDURJAL - 506212.

M.G.V.C ART'S COMMERCE AND SCIENCE COLLEGE, MUDDEBIHAL-586212

DEPARTMENT OF CHEMISTRY

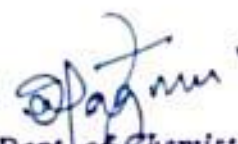
LIST OF SLOW AND ADVANCED LEARNERS

Class: B.Sc- I Semester

Year: 2024-25

Slow Learners		
SLNo.	Reg. No.	Name
1	U15NU24S0001	Mahant Malagimani
2	U15NU24S0031	Suresh C
3	U15NU24S0048	Tahreen Bagalkot
4	U15NU24S0062	Purushottam P Havaladar
5	U15NU24S0068	Deepa N Arabi
6	U15NU24S0069	Parashuram H Dharennavar
7	U15NU24S0077	Sidiya M Naikodi
8	U15NU24S0098	Rohit Pandappa Nayak
9	U15NU24S0104	Abdul latif K Nadaf
Advanced Learners		
1	U15NU24S0005	Srushti B Naduvnamath
2	U15NU24S0024	Shreekant R Kondaguli
3	U15NU24S0042	Roopa Mallikarjun Channa
4	U15NU24S0044	Jayalaxmi S Biradar
5	U15NU24S0070	Srushthi Siddanna Waggar
6	U15NU24S0108	Madhumati Shivappa Chouri

- ❖ 1. Slow and advanced learners are listed / Identified on the bases of PUC-II %
- ❖ 2. Students who secured below 70 % Marks are listed as slow learners
- ❖ 3. Students who secured above 95 % Marks are listed as advanced learners


Head, Dept. of Chemistry,
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MUDDEBIHAL - 586212.

M.G.V.C ART'S COMMERCE AND SCIENCE COLLEGE, MUDDEBIHAL-586212

DEPARTMENT OF CHEMISTRY

TIME-TABLE OF SLOW AND ADVANCED LEARNERS

Class: B.Sc- I Semester

Year: 2024-25

SLOW LEARNERS REMEDIAL CLASSES

Day	Time	Faculty
Thursday	4:00 – 5:00 PM	ASB
	5:00 – 5:45 PM	CSK
Friday	4:00 – 5:00 PM	ASB
	5:00 – 5:45 PM	CSK

ADVANCED LEARNERS ENRICHMENT CLASSES

Day	Time	Faculty
Thursday	4:00 – 5:00 PM	CSK
	5:00 – 5:45 PM	ASB


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M.G.V.C ART'S COMMERCE AND SCIENCE COLLEGE, MUDDEBIHAL-586212

DEPARTMENT OF CHEMISTRY

ATTENDANCE REPORT OF SLOW LEARNERS 2024-25

Faculty Name: Shri. A S Bagawan

Sl. No.	Reg.No.	Name of the Student	3/10	4/10	10/10	18/10	25/10	7/11	14/11	15/11	22/11	28/11
1	U15NU24S0001	Mahant Malagimani	1	2	3	4	5	6	7	8	9	10
2	U15NU24S0031	Suresh C	1	2	3	4	5	6	7	8	9	10
3	U15NU24S0048	Tahreen Bagalkot	1	2	3	4	5	6	7	8	9	10
4	U15NU24S0062	Purushottam P Havaladar	1	2	-	3	4	5	6	7	8	9
5	U15NU24S0068	Deepa N Arabi	1	2	3	4	5	6	7	8	9	10
6	U15NU24S0069	Parashuram H Dharennavar	1	2	3	4	5	6	7	8	9	10
7	U15NU24S0077	Sidiya M Naikodi	1	2	3	4	5	6	7	8	9	10
8	U15NU24S0098	Rohit Pandappa Nayak	1	2	3	4	5	6	7	8	9	10
9	U15NU24S0104	Abdul latif K Nadaf	1	2	3	4	5	6	7	8	9	10
Faculty Signature			<i>ASB</i>	<i>ASB</i>	<i>ASB</i>	<i>ASB</i>	<i>ASB</i>	<i>ASB</i>	<i>ASB</i>	<i>ASB</i>	<i>ASB</i>	<i>ASB</i>

ASB
Head, Dept. of Chemistry,
M.G.V.C. Arts, Com. & Science College,
MUDDEBIHAL - 586212.

M.G.V.C ART'S COMMERCE AND SCIENCE COLLEGE, MUDDEBIHAL-586212

DEPARTMENT OF CHEMISTRY

ATTENDANCE REPORT OF SLOW LEARNERS 2024-25

Faculty Name: Shri. C S Katageri

SL No.	Reg.No.	Name of the Student	$\frac{3}{10}$	$\frac{4}{10}$	$\frac{10}{10}$	$\frac{24}{10}$	$\frac{25}{10}$	$\frac{8}{11}$	$\frac{14}{11}$	$\frac{21}{11}$	$\frac{22}{11}$	$\frac{24}{11}$
1	U15NU24S0001	Mahant Malagimani	1	2	3	4	5	6	7	8	9	
2	U15NU24S0031	Suresh C	1	2	3	4	5	6	7	8	9	
3	U15NU24S0048	Tahreem Bagalkot	1	2	3	4	5	6	7	8	9	
4	U15NU24S0062	Purushottam P Havaladar	1	2	-	3	4	5	6	7	8	
5	U15NU24S0068	Deepa N Arabi	1	2	3	4	5	6	7	-	8	
6	U15NU24S0069	Parashuram H Dharennavar	1	2	3	4	5	6	7	8	9	
7	U15NU24S0077	Sidiya M Naikodi	1	2	3	-	4	5	6	7	8	
8	U15NU24S0098	Rohit Pandappa Nayak	1	2	3	4	5	6	7	8	9	
9	U15NU24S0104	Abdul latif K Nadaf	1	2	3	4	5	6	7	8	9	
Faculty Signature												

Test is conducted on 29.11.2024


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DEPARTMENT OF CHEMISTRY

ATTENDANCE REPORT OF ADVANCED LEARNERS 2024-25

Faculty Name: Shri. C S Katageri.

Sl. No.	Reg.No.	Name of the Student	<u>3</u> 10	<u>10</u> 10	<u>24</u> 10	<u>7</u> 11	<u>14</u> 11	<u>21</u> 11	<u>28</u> 11	<u>5</u> 12	<u>12</u> 12	<u>19</u> 12
1	U15NU24S0005	Srushti B Naduvinamath	1	2	3	4	5	6	7	8	9	
2	U15NU24S0024	Shreekant R Kondaguli	1	2	3	4	5	6	-	7	8	
3	U15NU24S0042	Roopa M Channa	1	2	3	4	5	6	7	8	9	
4	U15NU24S0044	Jayalaxmi S Biradar	1	2	3	4	5	-	6	7	8	
5	U15NU24S0070	Srushthi SWaggar	1	2	-	3	4	5	6	7	8	
6	U15NU24S0108	Madhumati SChouri	1	2	3	4	5	6	7	8	9	
Faculty Signature												


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DEPARTMENT OF CHEMISTRY

ATTENDANCE REPORT OF ADVANCED LEARNERS 2024-25

Faculty Name: Shri. A S Bagawan

Sl. No.	Reg.No.	Name of the Student	$\frac{3}{10}$	$\frac{10}{10}$	$\frac{24}{10}$	$\frac{7}{11}$	$\frac{14}{11}$	$\frac{21}{11}$	$\frac{28}{11}$	$\frac{5}{12}$	$\frac{12}{12}$	$\frac{19}{12}$
1	U15NU24S0005	Srushti B Naduvinamath	1	2	3	4	5	6	7	8	9	
2	U15NU24S0024	Shreekant R Kondaguli	1	2	3	4	5	6	-	7	8	
3	U15NU24S0042	Roopa M Channa	1	2	3	4	5	6	7	8	9	
4	U15NU24S0044	Jayalaxmi S Biradar	1	2	3	4	5	-	6	7	8	
5	U15NU24S0070	Srushthi SWaggar	1	2	-	3	4	6	7	8	9	
6	U15NU24S0108	Madhumati SChouri	1	2	3	4	5	6	7	8	9	
Faculty Signature			Feb	Feb	Feb	Feb	Feb	Feb	Feb	Feb	Feb	Feb

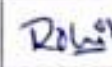
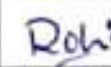
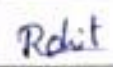
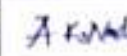
Test is Conducted on 19.12.2024


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DEPARTMENT OF CHEMISTRY

BOOKS AND STUDY MATERIALS FOR SLOW LEARNERS 2024-25

Sl. No.	Reg. No.	Name of the Student				
1	U15NU24S0001	Mahant Malagimani				
2	U15NU24S0031	Suresh C				
3	U15NU24S0048	Tahreen Bagalkot				
4	U15NU24S0062	Purushottam P Havaladar	P.P.Havaladar	P.P.Havaladar	P.P.Havaladar	P.P.Havaladar
5	U15NU24S0068	Deepa N Arabi				
6	U15NU24S0069	Parashuram H Dharennavar	P.H.D	P.H.D	P.H.D	P.H.D
7	U15NU24S0077	Sidiya M Naikodi				
8	U15NU24S0098	Rohit Pandappa Nayak				
9	U15NU24S0104	Abdul latif K Nadaf				

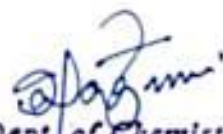

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DEPARTMENT OF CHEMISTRY

BOOKS AND STUDY MATERIALS FOR ADVANCED LEARNERS 2024-25

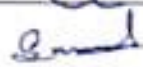
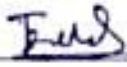
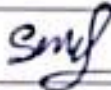
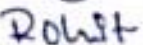
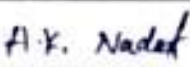
Sl. No.	Reg. No.	Name of the Student				
1	U15NU24S0005	Srushti B Naduvinamath	S.B.N.	S.B.N	S.B.N	S.B.N
2	U15NU24S0024	Shreekant R Kondaguli	S.R. Kondaguli	S.R. Kondaguli	S.R. Kondaguli	S.R. Kondaguli
3	U15NU24S0042	Roopa M Channa	R.M. Channa	R.M. Channa	R.M. Channa	R.M. Channa
4	U15NU24S0044	Jayalaxmi S Biradar	J.S.B.	J.S.B	J.S.B	J.S.B
5	U15NU24S0070	Srushthi SWaggar	S.S. Swaggar	S.S. Swaggar	S.S. Swaggar	S.S. Swaggar
6	U15NU24S0108	Madhumati Schouri	M.S.	M.S	M.S	M.S


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DEPARTMENT OF CHEMISTRY
TEST CONDUCTED FOR SLOW LEARNERS 2024-25

Date: 29.11.2024

Time: 5.00 - 5.45 PM

Sl.No.	Reg. No.	Name of the Student	Signature
1	U15NU24S0001	Mahant Malagimani	
2	U15NU24S0031	Suresh C	
3	U15NU24S0048	Tahreen Bagalkot	
4	U15NU24S0062	Purushottam P Havaladar	
5	U15NU24S0068	Deepa N Arabi	
6	U15NU24S0069	Parashuram H Dharenhavar	
7	U15NU24S0077	Sidiya M Naikodi	
8	U15NU24S0098	Rohit Pandappa Nayak	
9	U15NU24S0104	Abdul latif K Nadaf	


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M.G.V.C ART'S COMMERCE AND SCIENCE COLLEGE, MUDDEBIHAL-586212
DEPARTMENT OF CHEMISTRY
UNANNOUNCED TEST CONDUCTED FOR ADVANCED LEARNERS 2024-25

Date: 29/11/2025

Time:

Sl.No.	Reg. No.	Name of the Student	Signature
1	U15NU24S0005	Srushti B Naduvinamath	S.B.N
2	U15NU24S0024	Shreekant R Kondaguli	S.R. Kondaguli
3	U15NU24S0042	Roopa M Channa	R.M. Channa
4	U15NU24S0044	Jayalaxmi S Biradar	J.S.B
5	U15NU24S0070	Srushthi SWaggar	S.S. Waggar
6	U15NU24S0108	Madhumati SChouri	M.S.


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Class: B.Sc- I Semester

Year: 2024-25

Time: 1 Hour

Date: 25-10-2024

Maximum Marks: 20

Section A: Objective Type Questions (2 marks each)

(Answer any 5 questions)

1. Which of the following is a strong acid?
a) HCl b) H₂CO₃ c) CH₃COOH d) H₂O
2. What is the unit of the gas constant (R) in the ideal gas law equation?
a) atm·L/mol·K b) J/mol·K c) L·mol·K d) mol·L·K
3. Which of the following is a chemical property of sodium?
a) It is a good conductor of electricity b) It reacts with water to form sodium hydroxide
c) It is malleable d) It is ductile
4. What is the pH value of pure water?
a) 5 b) 7 c) 9 d) 14
5. Which of the following is a primary method of separating mixtures?
a) Evaporation b) Dialysis c) Chromatography d) All of the above

Section B: Long Answer Questions (5 marks each)

6. State Dalton's Atomic Theory. Discuss its significance and limitations.
7. Explain the ideal gas laws with the formula and conditions for their validity. Provide an example of how the ideal gas law is applied in real life.
8. What is Avogadro's hypothesis? Derive the relationship between the volume of a gas and the number of molecules at constant temperature and pressure.

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DEPARTMENT OF CHEMISTRY

ATTENDANCE REPORT OF SLOW LEARNERS I-IA

Class: B.Sc- I Semester

Year: 2024-25

Time: 1 Hour

Date: 25-10-2024

Sl. No.	Reg.No.	Name of the Student	Signature
1	U15NU24S0001	Mahant Malagimani	
2	U15NU24S0031	Suresh C	Suresh. C
3	U15NU24S0048	Tahreen Bagalkot	TBB
4	U15NU24S0062	Purushottam P Havaladar	P.P.Havaladar
5	U15NU24S0068	Deepa N Arabi	Deepa
6	U15NU24S0069	Parashuram H Dharennavar	P.H.D.
7	U15NU24S0077	Sidiya M Naikodi	Sma
8	U15NU24S0098	Rohit Pandappa Nayak	R.P. Nayak
9	U15NU24S0104	Abdul latif K Nadaf	Alkna


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DEPARTMENT OF CHEMISTRY

SLOW LEARNERS II-IA

Class: B.Sc- I Semester

Year: 2024-25

Time: 1 Hour

Date: 22-11-2024

Maximum Marks: 20

Section A: Very Short Answer Questions (1 mark each) ($5 \times 1 = 5$ marks).

1. Define atomic number.
2. What is the chemical formula of water?
3. Name one strong acid.
4. Write the symbol of Sodium.
5. What is the SI unit of mass?

Section B: Short Answer Questions (2 marks each) ($5 \times 2 = 10$ marks)

6. Differentiate between a compound and a mixture.
7. Write any two differences between an acid and a base.
8. What is the law of conservation of mass? Give one example.
9. Define isotopes. Give one example.
10. State any two characteristics of a chemical reaction.

Section C: Long Answer Question ($1 \times 5 = 5$ marks)

11. Explain the structure of an atom with a neat labelled diagram.

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DEPARTMENT OF CHEMISTRY

ATTENDANCE REPORT OF SLOW LEARNERS II-IA

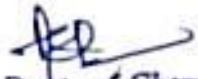
Class: B.Sc- I Semester

Year: 2024-25

Time: 1 Hour

Date: 22-11-2024

Sl. No.	Reg.No.	Name of the Student	Signature
1	U15NU24S0001	Mahant Malagimani	
2	U15NU24S0031	Suresh C	Suresh.C
3	U15NU24S0048	Tahreen Bagalkot	TB
4	U15NU24S0062	Purushottam P Hqvaldar	P.P.Havaldar
5	U15NU24S0068	Deepa N Arabi	Deepa
6	U15NU24S0069	Parashuram H Dharennavar	P.H.D.
7	U15NU24S0077	Sidiya M Naikodi	S.M.N.
8	U15NU24S0098	Rohit Pandappa Nayak	R.P. Nayak
9	U15NU24S0104	Abdul latif K Nadaf	A.K.Nadaf


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DEPARTMENT OF CHEMISTRY

RESULT SHEET OF SLOW LEARNERS I AND II-IA

Class: B.Sc- I Semester

Year: 2024-25

Sl. No.	Reg.No.	Name of the Student	I-IA	II-IA
1	U15NU24S0001	Mahant Malagimani	16	17
2	U15NU24S0031	Suresh C	17	18
3	U15NU24S0048	Tahreen Bagalkot	18	18
4	U15NU24S0062	Purushottam P Havaladar	19	17
5	U15NU24S0068	Deepa N Arabi	19	19
6	U15NU24S0069	Parashuram H Dharennavar	17	17
7	U15NU24S0077	Sidiya M Naikodi	18	18
8	U15NU24S0098	Rohit Pandappa Nayak	16	16
9	U15NU24S0104	Abdul latif K Nadaf	18	18


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I-1A		TOTAL MARKS OBTAINED
STUDENT'S NAME	Deepa Mittal	19/19
CLASS	B.Sc I	
ROLL NO.	1115N0245068	
DATE		25/10/2024

✓ A.1. a) ~~HCl~~

✓ 2. b) ~~J/mol.k~~

✓ 3. b) ~~It reacts with water to form sodium hydroxide.~~

✓ 4. b) ~~7~~

✓ 5. d) ~~All of the above~~

Q6. Dalton's "Atomic Theory":

* All matter is made of indivisible particles called atoms.

* Atoms of the same element are identical in mass and properties.

* Atoms of different elements have different masses and properties.

* Atoms combine in simple whole-number ratios to form compounds.

Significance: It provided a framework for understanding chemical reactions.

Nikita Silver

STUDENT'S NAME Deepa. Anbi		TOTAL MARKS OBTAINED 19/19
CLASS B.Sc I	SUBJECT Chemistry	
ROLL NO. 185AN245068	DATE 22/11/2024	

1. Atomic number is the number of protons in the nucleus of an atom.

2. H_2O

3. Hydrochloric acid (HCl)

4. ~~NaS~~

5. Kilogram (kg)

6. * A compound is made of two or more elements chemically combined (e.g., H_2O).

* A mixture contains two or more substances physically mixed, not chemically (e.g., air).

7. * Acids taste sour; bases taste bitter.

* Acids turn blue litmus red; bases turn red litmus blue.

8. * The law states that mass is neither created nor destroyed in a chemical reaction.

* Example: When hydrogen and oxygen form water, the total mass remains the same.

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DEPARTMENT OF CHEMISTRY

UNANNOUNCED TEST FOR ADVANCED LEARNERS I-IA

Class: B.Sc- I Semester

Year: 2024-25

Time: 1 Hour

Date: 07-11-2024

Maximum Marks: 20

Answer all questions. Each question carries 4 marks. ($5 \times 4 = 20$ marks)

1. Explain the concept of hybridization Describe the hybridization in Methane (CH_4).
2. Define and explain the Beer-Lambert Law in spectroscopy.
3. Describe the preparation and properties of Grignard reagents.
4. Write the electronic configuration of chromium and explain the exception to the Aufbau principle observed in it.
5. What is the difference between ionic and covalent bonds? Give one example of each.


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DEPARTMENT OF CHEMISTRY

UNANNOUNCED TEST FOR ADVANCED LEARNERS I-IA

Class: B.Sc- I Semester

Year: 2024-25

Time: 1 Hour

Date: 07-11-2024

Maximum Marks: 20

Sl. No.	Reg.No.	Name of the Student	Signature
1	U15NU2450005	Srushti B Naduvinamath	S.B.N
2	U15NU2450024	Shreekant R Kondaguli	Shk
3	U15NU2450042	Roopa M Channa	R.M.Channa
4	U15NU2450044	Jayalaxmi S Biradar	J.S. Biradar
5	U15NU2450070	Srushthi SWaggar	S.S. Swaggar
6	U15NU2450108	Madhumati SChouri	M.S.C


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DEPARTMENT OF CHEMISTRY

UNANNOUNCED TEST FOR ADVANCED LEARNERS-IIA

Class: B.Sc- I Semester

Year: 2024-25

Time: 1 Hour

Date: 19-12-2024

Maximum Marks: 20

Sl. No.	Reg.No.	Name of the Student	Signature
1	U15NU24S0005	Srushti B Naduvinamath	S.B.N.
2	U15NU24S0024	Shreekant R Kondaguli	SRK
3	U15NU24S0042	Roopa M Channa	R.M. Channa
4	U15NU24S0044	Jayalaxmi S Biradar	J.S. Biradar
5	U15NU24S0070	Srushthi SWaggar	Srushthi
6	U15NU24S0108	Madhumati SChouri	M.S.C.


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DEPARTMENT OF CHEMISTRY

UNANNOUNCED TEST FOR ADVANCED LEARNERS II-IA

Class: B.Sc- I Semester

Year: 2024-25

Time: 1 Hour

Date: 19-12-2024

Maximum Marks: 20

Section A: Short Answer Questions (2 marks each) ($5 \times 2 = 10$ marks)

1. Define the term molarity and write its unit.
2. Write the IUPAC name of $\text{CH}_3\text{-CH=CH-CH}_3$.
3. State Le Chatelier's Principle.
4. Write the electronic configuration of Fe (atomic number 26).
5. What is nucleophilic substitution reaction? Give one example.

Section B: Long Answer Questions (5 marks each) ($2 \times 5 = 10$ marks)

6. Derive the rate law expression for a first-order reaction and calculate the rate constant if the half-life is 20 minutes.
7. Explain the preparation and properties of benzene.
8. Describe the preparation of potassium permanganate (KMnO_4) by the oxidation of manganese dioxide.

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DEPARTMENT OF CHEMISTRY

RESULT SHEET FOR ADVANCED LEARNERS I AND I-IIA

Class: B.Sc- I Semester

Year: 2024-25

Sl. No.	Reg.No.	Name of the Student	I-IA	II-IA
1	U15NU24S0005	Srushti B Naduvinamath	19	19
2	U15NU24S0024	Shreekant R Kondaguli	19	20
3	U15NU24S0042	Roopa M Channa	20	19
4	U15NU24S0044	Jayalaxmi S Biradar	19	20
5	U15NU24S0070	Srushthi SWaggar	20	20
6	U15NU24S0108	Madhumati SChouri	20	20

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STUDENT'S NAME <u>Sushruti S. Waghare</u>		TOTAL MARKS OBTAINED <u>20</u> <u>20</u>
CLASS <u>B.Sc. I</u>	SUBJECT <u>Chemistry</u>	
ROLL NO. <u>1115N02952070</u>	DATE <u>07/11/24</u>	

1. Hybridization is the concept of mixing atomic orbitals to form new hybrid orbitals suitable for the pairing of electrons to form chemical bonds in molecules. In methane (CH_4), the carbon atom undergoes sp^3 hybridization where one s orbital and three p orbitals combine to form four equivalent sp^3 hybrid orbitals. These hybrid orbitals overlap with the $1s$ orbitals of hydrogen atoms, resulting in four sigma bonds arranged tetrahedrally.

2. Beer-Lambert Law relates the absorption of light to the properties of the material through which the light is traveling. It is expressed as: $A = \epsilon c l$, where A = absorbance, ϵ = molar absorptivity, l = path length, c = concentration. The law states that absorbance is directly proportional to the concentration of the absorbing species and the path length of the sample.

3. Grignard reagents are organomagnesium halides (RMgX) prepared by reacting alkyl or aryl halides with magnesium metal in dry ether.

Nikita Silver

STUDENT'S NAME <u>Shristhi S. Waghare</u>		TOTAL MARKS OBTAINED
CLASS <u>B.Sc I</u>	SUBJECT <u>Chemistry</u>	<u>20</u>
ROLL NO. <u>UJSH02450070</u>	DATE <u>19/12/24</u>	<u>20</u>

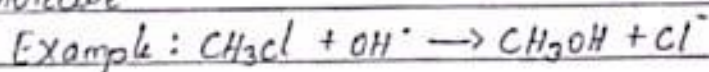
1. Molarity (M): Number of moles of solute dissolved per liter of solution. unit: moles/liter (mol/l).

2. IUPAC name: But-2-ene (since there is a double bond between carbons 2 and 3).

3. Le Chatelier's Principle: If a system at equilibrium is disturbed by a change in concentration, temperature, or pressure, the system shifts its equilibrium position to counteract the change.

4. Electronic configuration of Fe (Z=26): $1s^2 2s^2 2p^6 3s^2 3p^6 3d^6 4s^2$ or simplified as $[Ar] 3d^6 4s^2$.

5. Nucleophilic Substitution: A reaction in which a nucleophile replaces leaving group in a molecule.



6. Rate law for first-order reaction:

Rate = $k[A]$

Integrated form: $\ln[A] = -kt + \ln[A_0]$

Half-life, $t_{1/2} = 0.693/k$

Given $t_{1/2} = 20 \text{ min.}$

$k = 0.693/20 = 0.03465 \text{ min}^{-1}$

Nikita Silver